

Chapter-3

"Biomolecule"

- Biomolecules \rightarrow Carbon containing chemicals and molecules that are involved in metabolic processes of living organisms are called Biomolecule or Biological molecules.
- Are all living organisms made of same chemicals, i.e. element and compounds??
- On elemental analysis of plant tissue, animal tissue or microbial paste, we obtain a list of elements like C, H, O and many other Elements
- But when we perform same chemical/elemental analysis of on a piece of earth's crust (non-living matter) we obtain similar list.

* All elements present in earth's crust are also present in the sample of living tissue. [Qualitatively]

** On deep analysis or closer examination, scientists reveals that the relative abundance of C, H and O is higher in any living organism than in earth's crust.

\Rightarrow "The elemental composition of living tissues and non-living matter appear to be similar when analysed qualitatively"

TABLE 9.1 A Comparison of Elements Present in Non-living and Living Matter*

Element	% Weight of	
	Earth's crust	Human body
Hydrogen (H)	0.14	0.5
Carbon (C)	0.03	18.5
Oxygen (O)	46.6	65.0
Nitrogen (N)	very little	3.3
Sulphur (S)	0.03	0.3
Sodium (Na)	2.8	0.2
Calcium (Ca)	3.6	1.5
Magnesium (Mg)	2.1	0.1
Silicon (Si)	27.7	negligible

* Adapted from CNR Rao, Understanding Chemistry, Universities Press, Hyderabad.

9.1 How to analyse chemical composition \rightarrow

- we can find out chemical composition of living organism by performing chemical analysis.

- take living tissue (like a vegetable or a piece of liver) + Grind it with CH_3COOH (Trichloroacetic acid) using mortar and pestle

we obtain thick slurry

If we were to strain through

a cheesecloth or cotton

we obtain two fractions

Filtrate or acid Soluble pool

- contain thousands of organic compounds

It also contain

inorganic compounds and elements

• they have molecular

wt ranging from

18-800 daltons (Da)

• It roughly represent

cytoplasmic composition

Residue or

acid insoluble

fraction

It has four type

of organic compounds

1. Proteins

2. Nucleic acids

3. Polysaccharides

4. Lipids

\downarrow

These compounds are

separated from each

other by various technique.

Finally we isolate & purify

compound.

Analytical techniques,

are applied to find out

molecular formula &

probable structure of

compound.

mol. wt $\geq 10,000$ daltons (Da)

* Inorganic compounds or elements are best detected by destructive experiment.

Take a small amount of living

tissue & weigh it

\rightarrow The weight obtained is "wet weight"

\downarrow Dry it by process of heating

\downarrow \rightarrow All H_2O evaporates

we obtain dry weight

\downarrow Fully burn it (combustion)

\downarrow remaining part is called "Ash"

Carbon compounds converted to $\uparrow \text{CO}_2$ & H_2O

Ash contains inorganic elements like Ca^{2+} , Mg^{2+} , Na^+ , K^+

\therefore Elemental analysis gives elemental composition of

living tissue in the form of H, C, O, N, P etc.

while analysis for compounds gives an idea

of the kind of organic and in-organic constituents

present in living tissue.

Biomolecules

\downarrow

Biomolecule

mol wt < 1000 Da

\downarrow

Biomacromolecule

(found in acid insoluble fraction)

✗ Lipids are not polymer \Rightarrow mol. wt < 800 Da

they are of smaller molecular wt \therefore they are

not macromolecule.

✗ when we grind tissue, we are disrupting

the cell structure. cell membrane and other

membranes are broken into pieces, & form vesicles

which are insoluble in water.

\therefore These membrane fragments in form of vesicles

get separated along with the acid insoluble pool

and hence in macromolecular fraction.

chemical composition of living tissue

(average)

Component

% of the total cellular mass

Water

70-90%

Proteins

10-15%

Carbohydrate

3%

Lipids

2%

Nucleic acids

5-7%

Ions

1%

✗ water is most abundant chemical in living

organisms.